

Solution Quiz 3a: Modern Physics**Date: 12 April 2018****Useful Formulae**

$$h = 6.63 \times 10^{-34} \text{ Js} \qquad m_e = 9.11 \times 10^{-31} \text{ kg}$$

$$E_n = \frac{-13.6}{n^2} \text{ eV in a } H \text{ atom} \qquad 1 \text{ eV} = 1.6 \times 10^{-19} \text{ J.}$$

1. The radial wavefunction for $n = 2$, $\ell = 1$ is given by,

$$R(r) = A \left(\frac{r}{a_0} \right) e^{-(r/2a_0)},$$

where A is a constant and a_0 is the Bohr's radius. Where is the maximum likelihood of finding the electron in this state? (5 marks)

Answer 1

Radial probability density $p_r(r)$ is

$$p_r(r) = 4\pi r^2 |R(r)|^2$$

$$= \frac{4\pi A^2}{a_0^2} r^4 e^{-r/a_0}$$

For finding the r at which this density is maximum,

$$\frac{dp_r(r)}{dr} = \frac{4\pi A^2}{a_0^2} (4r^3 e^{-r/a_0} - \frac{1}{a_0} r^4 e^{-r/a_0}) = 0$$

$$\Rightarrow 4r^3 - \frac{1}{a_0} r^4 = 0$$

$$\Rightarrow r = 4a_0.$$

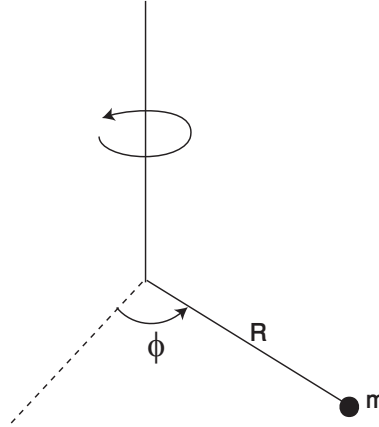
The maximum likelihood is at $r = 4a_0$.

2. A particle of mass m is at a fixed radius R from the origin. The moment of inertia is $I = mR^2$. The time dependent Schrodinger equation is,

$$-\frac{\hbar^2}{2I} \frac{\partial^2 \Psi(\phi, t)}{\partial \phi^2} = i\hbar \frac{\partial \Psi(\phi, t)}{\partial t},$$

where ϕ is the variable angle in space. Use separation of variable to

- write down the spatial (space) and temporal(time) component of the schrodinger equation. (5 marks)
- Solve the spatial part to find $\Phi(\phi)$. (5 marks)
- We want the spatial part $\Phi(\phi)$ to be single-valued meaning that if ϕ changes by 2π or multiples thereof, the function does not change, i.e., $\Phi(\phi + 2\pi n) = \Phi(\phi)$. What kind of quantization does this led to? Comment. (5 marks)



Answer 2

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$$\Psi(\phi, t) = \Phi(\phi) f(t)$$

Inserting into the L.H.S. and R.H.S.

$$\begin{aligned} -\frac{\hbar^2}{2I} \frac{\partial^2}{\partial \phi^2} (\Phi(\phi) f(t)) &= i\hbar \frac{\partial}{\partial t} (\Phi(\phi) f(t)) \\ -\frac{\hbar^2}{2I} \left(\frac{\partial^2 \Phi(\phi)}{\partial \phi^2} \right) f(t) &= i\hbar \Phi(\phi) \frac{\partial}{\partial t} f(t) \end{aligned}$$

Divide by $\Phi(\phi) f(t)$:

$$-\frac{\hbar^2}{2I} \frac{\partial^2 \Phi(\phi)}{\partial \phi^2} \frac{1}{\Phi(\phi)} = i\hbar \frac{1}{f(t)} \frac{\partial}{\partial t} f(t)$$

Each side is equal to a constant, say E . Hence spatial part is

$$\frac{d^2\Phi(\phi)}{d\phi^2} + \frac{2IE}{\hbar^2}\Phi(\phi) = 0$$

and the temporal part is

$$\frac{\partial}{\partial t}f(t) = -\frac{i}{\hbar} Ef(t).$$

(b) The spatial part is:

$$\frac{d^2\Phi(\phi)}{d\phi^2} + \frac{2IE}{\hbar^2}\Phi(\phi) = 0$$

The auxiliary equation is:

$$s^2 + \left(\frac{2IE}{\hbar^2}\right) = 0$$

$$s = \pm \frac{i\sqrt{2IE}}{\hbar}$$

Hence the solution is:

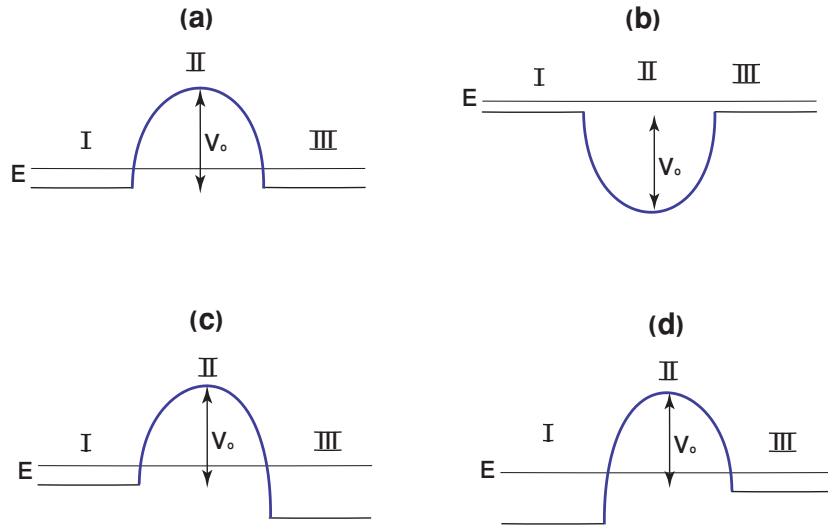
$$\Phi(\phi) = Ae^{+i\frac{\sqrt{2IE}}{\hbar}\phi} + Be^{-i\frac{\sqrt{2IE}}{\hbar}\phi}. \quad (1)$$

(c) The boundary condition is:

$$\begin{aligned} \Phi(\phi + 2\pi n) &= \Phi(\phi) \\ \Phi(\phi + 2\pi n) &= Ae^{+i\frac{\sqrt{2IE}}{\hbar}(\phi+2\pi n)} + Be^{-i\frac{\sqrt{2IE}}{\hbar}(\phi+2\pi n)} \\ &= Ae^{+i\frac{\sqrt{2IE}}{\hbar}\phi} e^{i\frac{\sqrt{2IE}}{\hbar}2\pi n} + Be^{-i\frac{\sqrt{2IE}}{\hbar}\phi} e^{-i\frac{\sqrt{2IE}}{\hbar}2\pi n} \end{aligned} \quad (2)$$

Equations (1) and (2) are equal requiring that $2\pi\frac{\sqrt{2IE}}{\hbar}n$ is an integral multiple of 2π since $e^{\pm i2\pi p} = 1$ where p is an integer. Hence we require that $\frac{\sqrt{2IE}}{\hbar}$ is an integer, which is the required quantization condition.

3. An electron is injected into a potential energy landscape from the left region I as shown below. It encounters a potential step. The energy of the electron is E and $E < |V_0|$. If the electron is to emerge in region III with a faster speed, the appropriate potential step is given by which of the following?



(e) The speed of the electron cannot increase.

Answer 3

In region III, we need v (speed) to be higher, so momentum ($p = \hbar k$) needs to be higher. A higher k is a shorter wavelength so, $\sqrt{E - V}$ should be higher in region III than in region I. This is the scenario in option (c).