Optimized synthesis of dichalcogenide films via chemical vapor deposition

SPROJ Final Presentation





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Contents

- Research Map
- TMDCs
- MoS₂
- Apparatus
- Stepwise synthesis
- Optimizing thickness
- Results
- Conclusion
- Future plans





What are transition metal dichalcogenides?

- What are TMDCs?
- Chemical formula = MX_2
- Semiconductors ~ reduced to nanostructures/2D materials
- Nanoscale materials vs bulk materials
- Created via exfoliation, physical or chemical vapor deposition (CVD), sputtering.
- Applications of TMDCs: catalysis, sensor applications, electronic devices, electrochemical reactions, energy storage, fuel cell, and renewable energy technology.



Why MoS₂?

Graphene

- Monolayers discovered in 2004; Nobel Prize in 2010
- Carbon atoms ~ hexagonal lattice
- Mechanical stiffness, flexibility, thermal and electric properties
- Zero bandgap No optical switching!



- It behaves similarly to graphene due to its structure and has a tunable bandgap so is a great substitute.
- Allows switching.
- The bandgap is layer dependent.





(Sanchez, Hummer, et al. 2015)

Properties and applications of MoS₂

- Shows n type and p type behavior depending on which dopant is used.
- Highly photosensitive and photo responsive due to the direct bandgap in nanoform.
- Recently being explored for optical and electric sensing applications.



(C. Vidya, et al., 2022)

Chemical vapor deposition (CVD) technique

The apparatus set up and basic growth procedure



Actual apparatus setup in Spin Physics Lab





Thermal Mapping of the furnace

• How do we know where to place precursors and substrates in the tube?



Visual observations of apparatus

outlet

Substrates and precursor before growth





Substrates and precursor after growth





inlet

Characterization Techniques

Optical microscopy

- To observe if growth has occurred
- Optical microscopy is performed at 50x magnification.





Raman Spectroscopy

- Noninvasive, nondestructive, vibrational spectroscopy technique.
- Helps identify chemical nature through the scattering produced by the in-plane and out of plane vibrations between the bonds.
- To identify the exact molecule, as the bonds are unique to each chemical makeup.
- MoS₂ has signature peaks E¹_{2g} and A_{1g}



⁽Tummala, Lamperti et al. 2020)

Scanning electron microscopy

Topography	 Appearance and surface features like texture
Crystallography	 How the atoms or flakes are arranged in the film
Morphology	 The size of the particles produced
Composition	 The elements that have played part in making the film and their respective compositions and ratio

Energy Dispersive X-ray Spectroscopy (EDX)



- Used with SEM to analyze the chemical composition of materials.
- A focused beam of electrons is scanned over the surface to produce an image, triggering and detecting X-ray emissions.
- These correspond to atomic levels present in sample, used to identify chemical composition.
- Provides detailed information on elemental composition, distribution, and structure.



Synthesis of MoS₂ thin films using a step-wise approach

Two-step growth of MoS₂ thin film



The schematic above shows the step-wise process to produce a thin film of MoS_2 along with the displacement over which the tube is physically moved.

(Wang, W., *et al.*, 2018)

Data obtained from two-step growth



M10

500nm

mode det spot

Optimization of parameters for two-step growth

	M10	M12	M10 (sulfurized)	M12 (sulfurized)
Substrate	Si	Si	Si	Si
MoO ₃ (mg)	35	25	0	0
S (mg)	1.5	1.5	3	3
Set temperature (°C)	800	800	800	800
Growth temperature (°C)	799	799	750	750
Gas	Ar	Ar	Ar	Ar
Working pressure (torr)	1.8	1.8	1.8	1.8
Gas flow (sccm)	400-150	400-50	100-70	200-100
Growth time (min)	30	30	30	30

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First target achieved!

Next, we challenged ourselves to produce MoS_2 in one step for efficiency.

One-step growth of MoS₂ thin film



 Only the parameters were slightly altered, the process remained the same to produce the twostep method results.

(Wang, W., *et al.*, 2018)

Data obtained from one step growth





Optimization of parameters for one-step growth

	M14	M15
Substrate	Si	Si
MoO ₃ (mg)	26	26
S (g)	6	6
Set temperature (°C)	800	800
Growth temperature (°C)	810	810
Gas	Ar	Ar
Working pressure (torr)	1.8	1.8
Gas flow (sccm)	100-70	100-70
Growth time (min)	30	30



- MoS₂ thin films were produced multiple times using this set of parameters.
- It was found to be more efficient as lesser energy, precursors and time were utilized.

Optimization of parameters to reduce the number of layers grown

Mass of precursor, position and orientation of the substrate

	H1, H2	H3, H4	H5, H6	H7,8,9		
Substrate	Si	Si	Si	Si		
MoO ₃ (mg)	13.8	8.3	5	7		
S (g)	6	4	4	6		
Set temperature (°C)	800	800	800	800		
Growth temperature	832.6	832.6	832.6	825.3		
(°C)						
Gas	Ar	Ar	Ar	Ar		
Working pressure	2.1	2.1	2.1	2.1		
(torr)						
Gas flow (sccm)	100	100	100	100		
Growth time (mins)	15	15	7	10		

- Reduced mass of precursors.
- Position and orientation varied.

Effect of precursor to substrate distance



Raman analysis for varying distances from the precursor



First goal of SPROJ B achieved!

- Peak separation 27cm⁻¹ represents bulk, and 25.3cm⁻¹ represents <u>5</u> <u>atomic layers</u>, we see this gradient of atomic layers along the length of sample H5.
- As we move further away from the precursor, we see a decrease in peak separation.
- This allows us to conclude that the position of the substrate and its distance from the precursor is vital to the growth of atomic layers.

Effect of substrate Orientation





Raman analysis for the change in orientation

- This sample was kept in standing.
- From our reference Raman plot, a peak difference of 24 cm⁻¹ corresponds to the presence of <u>4 atomic</u> <u>layers.</u>
- First optimization set allowed us to realize that <u>distance and orientation</u> of the substrate are vital to the growth of atomic layers.

Alkali metal assisted growth of MoS₂ films

Purpose of using alkali metals in assisting growth

	H19	H30,31,32	H33,34,35		
Substrate	Si	Si	Si		
MoO ₃ (mg)	10	5	5		
NaCl	15	15	15		
S (g)	2.00	2.00	2.00		
Set temperature (°C)	800	800	800		
Growth temperature (°C)	800	800	832.6		
Gas	Ar	Ar	Ar		
Working pressure (torr)	2.1	2.1	2.1		
Gas flow (sccm)	100	100	100		
Growth time (mins)	10	10	10		

- NaCl used: cost effective catalyst.
- Reduces temperature required for growth.
- NaCl changes morphology (can enhance it but may also drastically affect it).

Optical Images of NaCI assisted growth

Increasing temperature and increasing NaCI to MoS₂ ratio.

Low NaCI : MoS₂ ratio

High NaCI : MoS₂ ratio

Raman analysis of NaCl assisted growth

- Using NaCl has brought down the thickness towards <u>3 atomic layers</u>.
- This can be confirmed by the peak separation of 23.25 cm⁻¹ from reference.
- NaCl also improved the growth time by decreasing it.

SEM images of films produced with NaCl

• The image shows the film produced with triangular shapes or overlapping triangles.

(Wang, Rong et Al. 2014)

- To the left is a field of vertically standing triangular flakes of MoS_{2.}
- This is in line with literature which claims that the addition of NaCl alters the morphology.

(Jian, Chang, Xu. 2018)

Creating an alternative to a double heating zone furnace

Apparatus altered to create a double zone furnace

- The tube's end flanges were replaced with a cork and teflon tape combination, this was carefully secured in place.
- These changes allowed us more movement in the tube displacement, which allowed us to create a doublezone like furnace.

Altered apparatus

Altered apparatus

	H15	H1	H17,18	H20,H21	H22	H23	H24	H25	H26,27	H28,29
		6								
Substrate	Si	Si	Si	Si	Si	Si	Si	Si	Si	Si
MoO ₃ (mg)	5.00	4.90	2.50	5.00	5.00	5.00	3	0.5	10	8
S (g)	4.00	1.15	2.00	0.4	0.8	0.8	0.8	0.8	1.6	6
Set	800	800	800	800	800	700	800	800	800	800
temperature										
(°C)										
Growth	832.6	832.	832.6	832.6	832.6	760	832.6	832.6	832.6	832.6
temperature		6								
(°C)										
Gas	Ar	Ar	Ar	Ar	Ar	Ar	Ar	Ar	Ar	Ar
Working	2.1	2.1	2.1	2.1	2.1	2.1	2.1	2.1	2.1	2.1
pressure										
(torr)										
Gas flow	100	100	100	100-200	100-	100	100	200-	235	100
(sccm)					200			235		
Growth time	25	7	10	10	7	15	7	7	7	10
(mins)										

- The precursors were further decreased.
- The displacement of the tube was varied greatly by changes to the apparatus.
- Gas pressure was varied.
- Position of the substrate with respect to the precursor was varied.

Optical Images

- The images are instantly cleaner and sharper once the double zone is created.
- Rods, squares and quadrilateral shapes are visible with clear outlines.

SEM images and more clues!

200 nm

- Rod-like structures and quadrilaterals are visible.
- There appear to be flakes in the background and the foreground.
- The images in the bottom row were taken from a paper that produced MoS₂ the same way as us.

(John, Dhara.)

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(Jian, Xang, Ju. 2019.)

Towards a continuous monolayer

(Jian, Xang, Ju. 2019.)

- The image shows our results for this monolayer which appear to be a continuous film with a few flakes in the foreground
- The reference image for a monolayer taken from the literature shows an almost similar image of a continuous film!

A noisy monolayer!

 When deposition on the film is very less, it's detected signal from Raman can become very weak in strength.

These peaks appeared clearly in the Raman data but became very noisy once plotted.

 <u>The peak separation of</u> <u>17.39 cm⁻¹ indicates the</u> <u>presence of a monolayer.</u>

Best results plotted!

(Tummala, Lamperti et al. 2020)

Conclusion and future remarks

Our results

- The reference image shows the effect of temperature and pressure.
- Our results to the right ٠ show similarity to the reference image.

Acknowledgements

- Dr. Ata ul Haq, Dr. Ammar Ahmed, and their PhD students for their contribution to Raman spectroscopy
- Dr. Syed Adnan Raza
- PhysLab members
- Wardah Mahmood
- Dr. Sabieh Anwar

Thank you!

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