Name :			
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Question 1

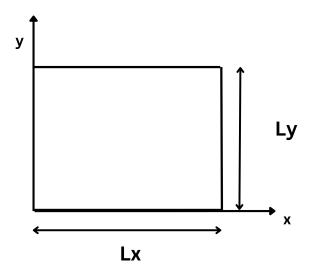
A hydrogen atom is in the state

$$\frac{1}{\sqrt{2}}(\psi_{100} - \psi_{200}).$$

Write the integral that, when computed, will give the probability of finding the electron between the radii r = 0 and $r = a_0$ (the Bohr radius). You do not have to compute this integral.

Question 2

Consider a particle in a 2-d potential well. The potential V(x,y) is infinite at the boundary and everywhere outside the well and zero inside the well. The length of the 2-d well is L_x and L_y as shown in the diagram.



The two dimensional Schrodinger equation for the particle is the following:

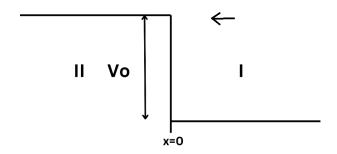
$$\frac{-\hbar^2}{2m} \left[\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} \right] \psi(x,y) + V(x,y)\psi(x,y) = E\psi(x,y).$$

- (a) Find a general expression for the possible energies of the given system. Also find the general expression for the corresponding wavefunctions.
 - It is possible that $L_x = L_y$ or $L_x \neq L_y$. Do parts b,c twice: once with the assumption that $L_x = L_y$ and once with the assumption that $L_x < L_y$.

- (b) Are energy levels degenerate for the first excited state? Justify your answer.
- (c) Write down the wavefunctions for the first excited state for each case you mentioned above.

Question 3

Consider the potential step shown below.



- (a) If a particle comes in from the right (as shown with the arrow) with $E > V_O > 0$, find its wavefunction in region I and II.
- (b) Use boundary conditions to write down all the possible coefficients of the wavefunctions we found in part (a).
- (c) What is the probability density of position in region I?

$$\frac{1}{\sqrt{100}} = \frac{1}{\sqrt{100}} = \frac{1$$

90 6 - 40 box

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$$\frac{-h^{2}}{am} \left[\frac{\partial}{\partial x^{2}} + \frac{\partial^{2}}{\partial y^{2}} \right] \Upsilon(x,y) + V(x,y) \Upsilon(x,y) = F \Upsilon(x,y)$$

$$\Psi(x,y) = XY \quad (\text{ceparation of variables}$$

$$\frac{-h^{2}}{dm} \left[Y \frac{\partial}{\partial x^{2}} X + X \frac{\partial}{\partial y^{2}} \right] + V X Y = F X Y$$

$$\text{divide both sides of equation by } XY$$

$$\frac{-h^{2}}{dm} \frac{1}{X} \frac{\partial}{\partial x^{2}} X + \frac{1}{Y} \frac{\partial}{\partial y^{2}} = (F - V).$$

$$\frac{-h^{2}}{dm} \frac{1}{X} \frac{\partial}{\partial x^{2}} X = F X \qquad \text{and} \quad \frac{1}{Y} \frac{\partial}{\partial y} Y = F y.$$

$$\text{For a } \partial D \text{ potential } V_{X=0} \text{ and } Y_{q=0} \text{ inside the box.}$$

$$\text{Salving } \frac{-h^{2}}{dm} \frac{\partial}{\partial x^{2}} X = X F x \qquad \Rightarrow \frac{\partial G}{\partial x} \frac{\partial^{2} X}{\partial x^{2}} = \frac{\partial m}{h^{2}} F x X$$

$$\text{Ansatz } X(x) = \frac{\partial m}{h^{2}} F x$$

Similarly, we can salue for
$$y$$
, $Ky^2 = \frac{2m}{\hbar^2} Ey$

$$Y(y) = C \sin(kx) + D \cos(kx).$$

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Boundary Conditions

$$X(0) = A \sin(0) + B \cos(0) = 0$$

$$= B = 0$$

$$X(Lx) = A \sin(L_x x_x) = 0$$

For a non-trivial solution
$$K_x L_x = n \times \text{ where } n = 1, 2, 3...$$

$$\sqrt{2mE_x} L_x = n \times \text{ where } n = 1, 2, 3...$$

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$$\sqrt{2mE_x$$

$$Y(y) = \left[\frac{2}{Ly} \sin\left(\frac{n \times y}{Ly}\right)\right]$$

$$E_y = \frac{n_y^2 \times h^2}{Ly^2 am}$$

$$\overline{E} = \frac{nx^2 x^2 h^2}{L^2 dm} + \frac{ny^2 x^2 h^2}{L^2 dm} = \frac{x^2 h^2}{L^2 dm} \left(nx^2 + ny^2 \right)$$

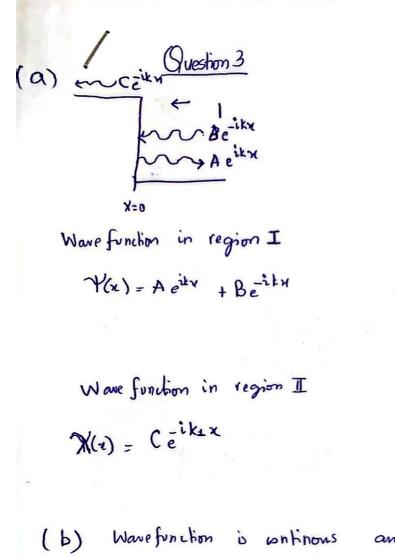
For the 1st excited state nx=1 and ny=2 OR

nx=2 and ny=1 F Degeneracy=

$$E = \frac{nx^2 x^2 k^2}{Lx^2 2m} + \frac{ny^2 x^2 k^2}{Ly^2 2m}$$

$$\gamma = \frac{2}{\sqrt{2}} \sin\left(\frac{n\pi\gamma}{L}\right) \sin\left(\frac{n\pi\chi}{L}\right)$$

$$\frac{dii}{\sqrt{2x} Ly} = \frac{2}{\sqrt{2x} Ly} \sin\left(\frac{n\pi y}{Ly}\right) \sin\left(\frac{n\pi x}{Lx}\right)$$



(b) Wavefunction is antinous and its desirative is also constant

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where Ki = Jam(E-V.)